

NAA at NC State

H. O. Back –Physics

S. Lassell – Nuclear Engineering

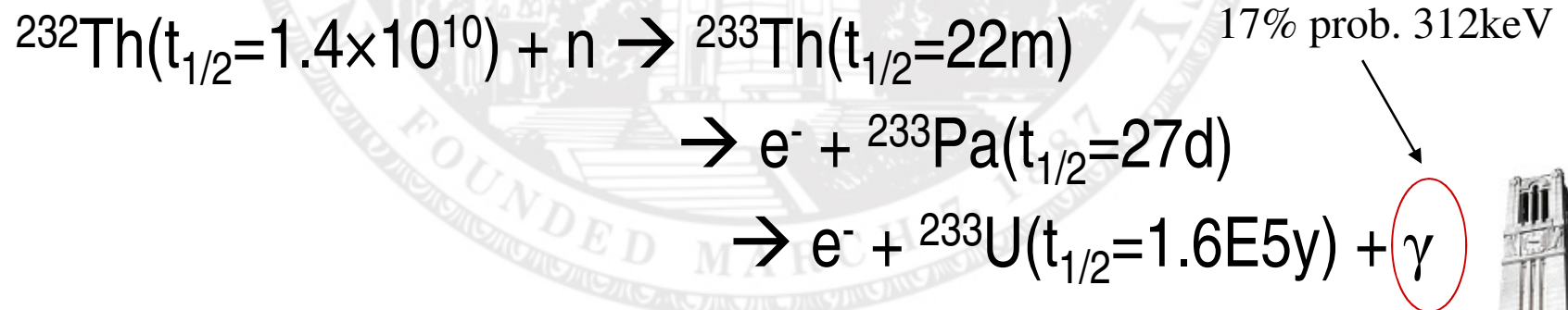
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NAA principle for U and Th

- Transmute long lived isotope into a short lived radioactive isotope through neutron activation
- Search for new isotope decay by identifying signature gammas through gamma ray spectroscopy
- Example:



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- Activate sample with high flux of neutrons and search for signature gammas to identify contaminant

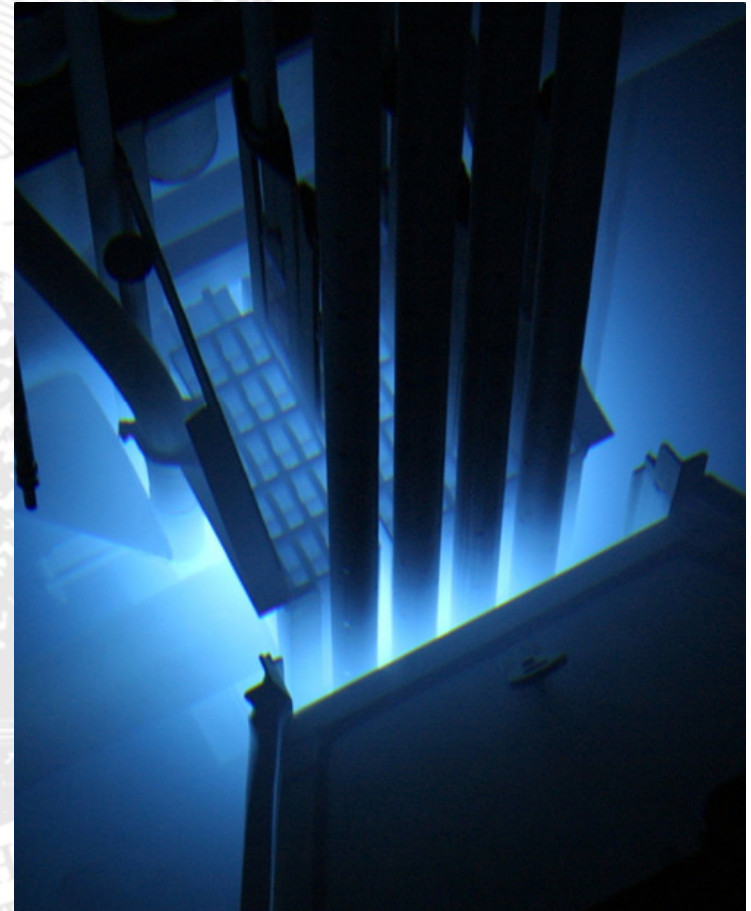
Element	Parent Isotope	Product Isotope and Decay chain	Gamma energy (keV)	Sensitivity
K	^{41}K	^{42}K ($t_{1/2}=12.26$ hr)	1524	Best limit 50 ppb Typical 50 ppb
U	^{238}U	^{239}U ($t_{1/2} = 24$ min) \rightarrow ^{239}Np ($t_{1/2} = 2.35\text{d}$)	106, 222, 228, 277	Best limit 1ppt Typical 20 ppt
Th	^{232}Th	^{233}Th ($t_{1/2} = 22$ min) \rightarrow ^{233}Pa ($t_{1/2} = 27\text{d}$)	312	Best limit 1 ppt Typical 20 ppt



Pulstar reactor

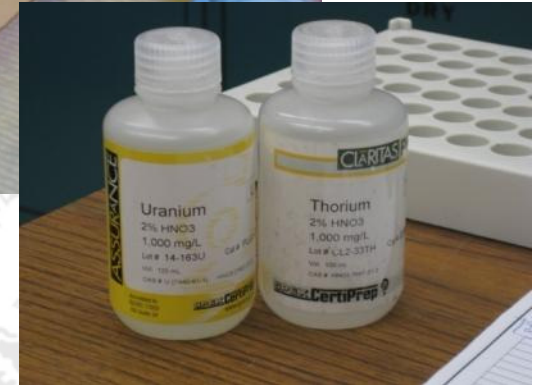
(http://www.ne.ncsu.edu/NRP/reactor_program.html)

- Educational reactor
- Provides services for basic research
- 1 MW pool type reactor
- 4% enriched uranium dioxide fuel
- Thermal n-flux in 'Rotating Exposure Ports' = $4-8 \times 10^{12}$ NV at 1 MW



NAA method at NC State

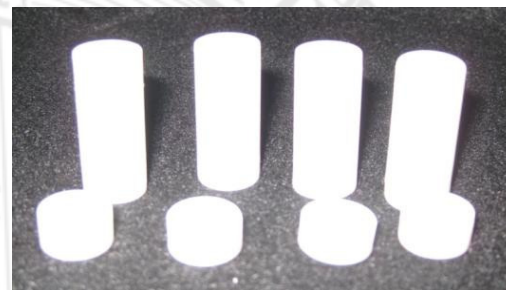
- Samples
 - Sealed in plastic vials
 - Vials are LDPE (tends to be cleaner)
- Standards
 - Aqueous solution sealed in plastic vials (vials are HDPE, which holds up better to irradiation)
 - Known amount of U and Th
 - Irradiated along with samples
 - Similar geometry as sample
- Flux monitor
 - Antimony solution sealed in plastic vials
 - Irradiated along with samples
 - Used to monitor n-flux if necessary
- All vials are packed and sealed in irradiation bottles (nalgene bottle)



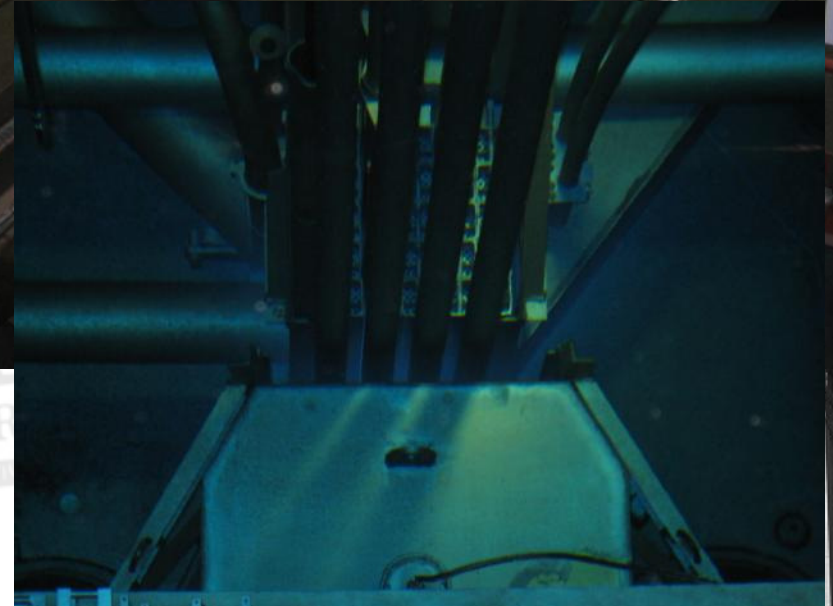
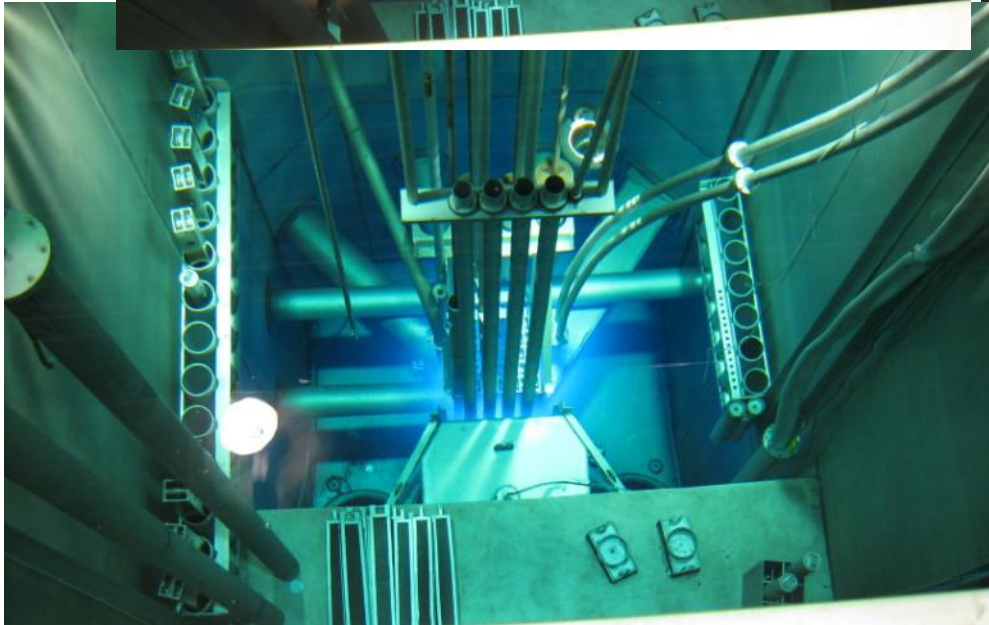
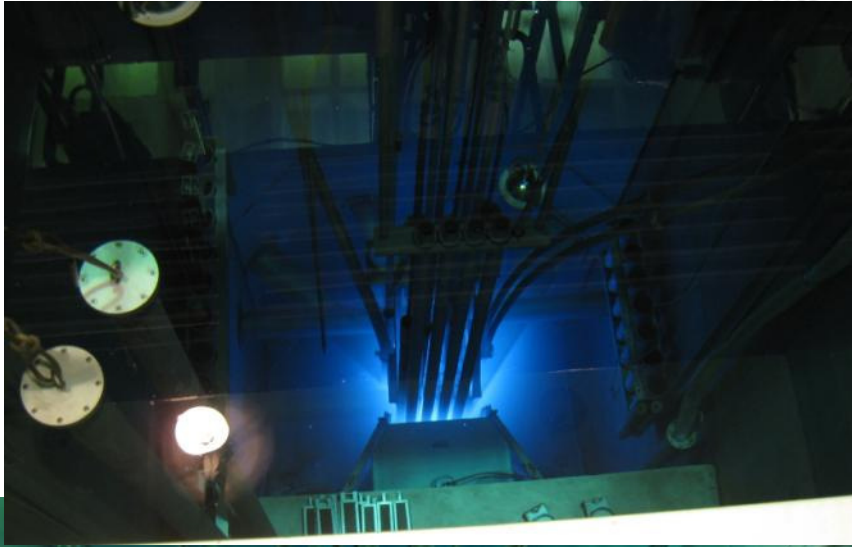
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Sample and Standard prep

- Sample
 - Plastic = TE-6472
 - Identified by EXO collaboration
 - Used for MAJORANA crystal mounting
 - Machined at NC State to fill vial radially and 2/3 of the vial height
 - Mass – 38.8g
 - Plastic and vials etched and cleaned by Eric Hoppe –PNNL
 - Vial heat sealed at NC State
- Standard
 - U and Th in 2% HNO_3 aqueous solution
 - Height of standard liquid equals height of sample to match geometries
 - Starts as 1000mg/L, but further diluted to give required mass with correct geometry
 - Total U mass = 50 μg
 - Total Th mass = 2.5 μg

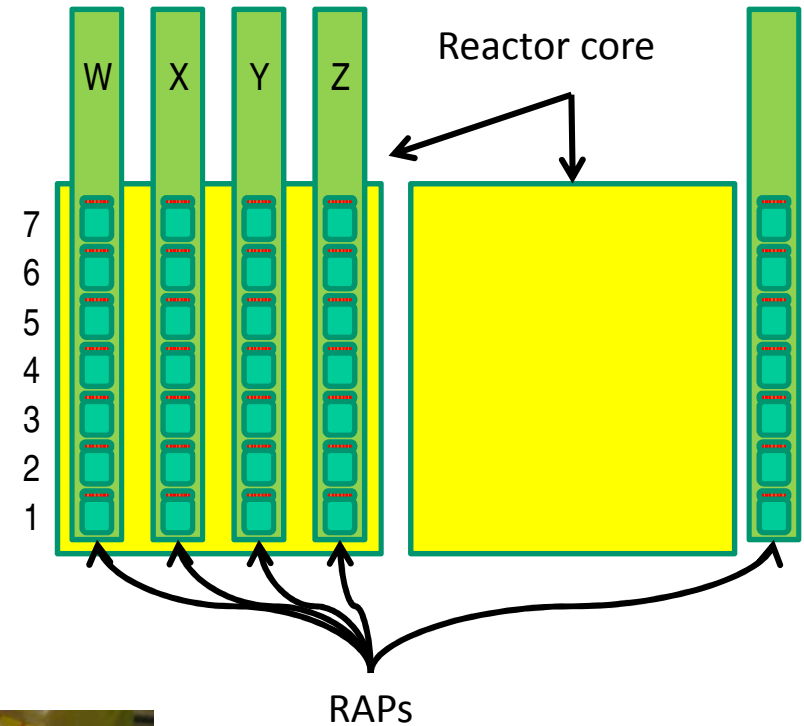


Reactor pictures



Irradiation

- Rotating Exposure Ports (RAPs)
 - Thermal n-flux in RAPs = $4-8 \times 10^{12}$ NV at 1 MW (the flux varies by position)
 - Four RAPs each hold 7 irradiation bottles
 - Irradiation bottles suspended in aluminum stringers
 - Stringers must 'cool' in reactor pool for 4+ days before irradiation bottles can be removed
 - Eliminates ^{40}K determination (^{42}K $t_{1/2} \sim 12\text{hrs}$)
 - Reduces ^{238}U determination efficiency (^{239}Np $t_{1/2} = 2.35\text{d}$)
- Our sample irradiation
 - Irradiation positions
 - Sample – 2Y (highest flux region)
 - U standard – 1Y
 - Th standard – 3Y
 - 18 MW hours



Post irradiation sample prep

- Samples removed from irradiated vials
- Sonicated in dilute HNO_3 solution for 1 hour
- Rinsed with DI H_2O in ultrasonic bath for 1 hour
- Samples placed in new, non-irradiated plastic vials to maintain geometry



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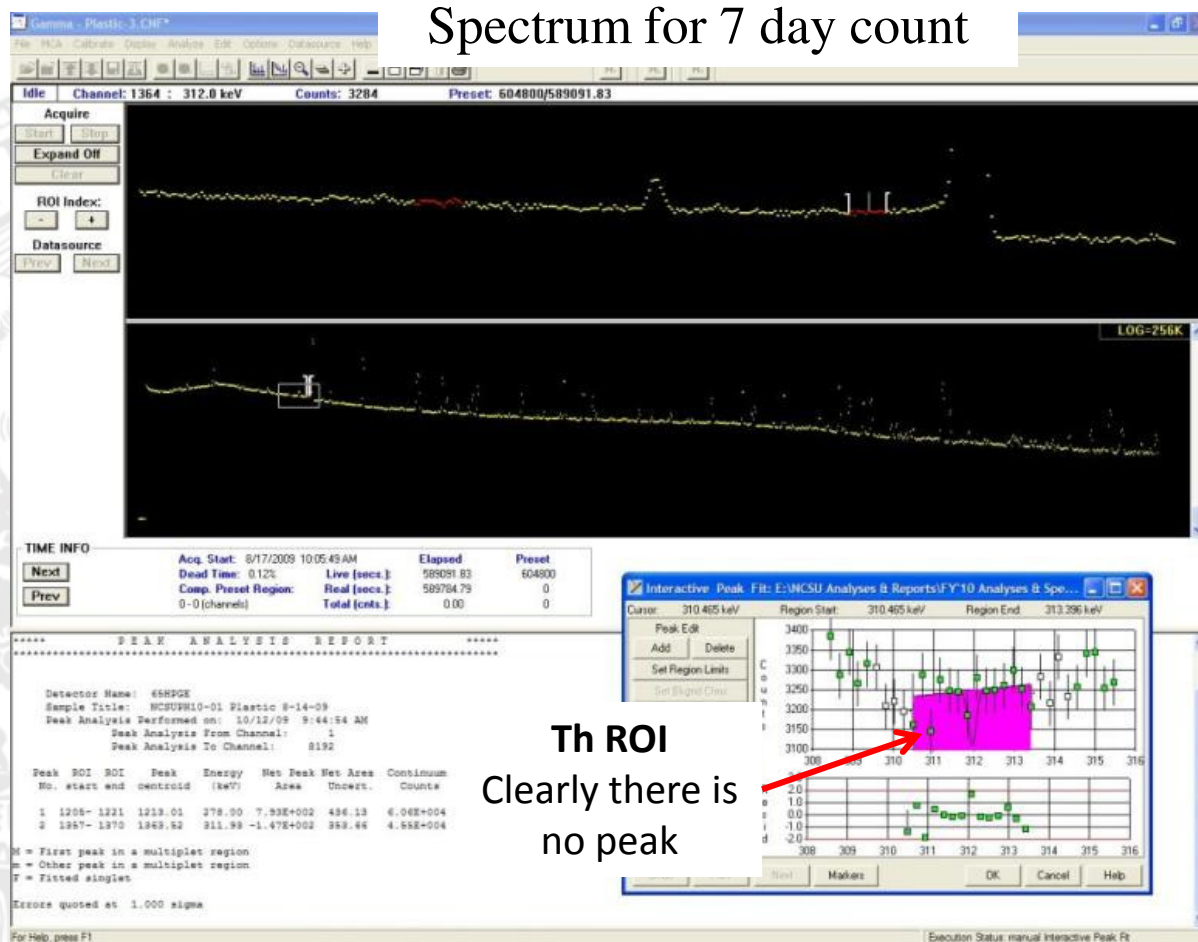
Counting

- Detector
 - 65% HPGe
 - 8" Pb shield
 - 2" poly shield
- Counting
 - All four vials counted together
 - Geometry of vials kept the same in each counting
 - Total counting campaign > 10 days



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Spectrum for 7 day count



Date		Start time	Counting time (sec)
8/11/09	Sample	3:15:54 pm	78,767.3
8/12/09	Thorium standard	1:46:29 pm	636.1
8/12/09	Sample	2:08:05 pm	86,577.4
8/17/09	Sample	10:05:49 am	589,784.8
8/21/09	Uranium standard	3:35:59 pm	371.9

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Results

- Analysis
 - Standards give expected gamma rate for given U and Th mass
 - Matching counting geometries cancel any geometrical systematic errors

- **Results**

- $^{232}\text{Th} \leq 0.809 \times 10^{-6} \text{ } \mu\text{g/g}$ or $3.28 \text{ } \mu\text{Bq/kg}$
 - $^{238}\text{U} \leq 2.77 \times 10^{-6} \text{ } \mu\text{g/g}$ or $34.5 \mu\text{Bq/kg}$

- Comparison – Best EXO results for TE-6472
 - $^{232}\text{Th} \leq 0.26 \times 10^{-6} \text{ } \mu\text{g/g}$
 - $^{238}\text{U} \leq 0.78 \times 10^{-6} \text{ } \mu\text{g/g}$

Systematic study of trace radioactive. D. S. Leonard, et Al. 2008, Nuclear Instruments and Method A 591, pp. 490-509.



Summary

- NAA at NC State is a viable material assay method for DUSEL experiments
 - NC State Nuclear Engineering is very supportive (private conversation with director of reactor program)
 - Is done completely in-house at NC State (reliably)
 - Irradiation (others irradiate at other facilities)
 - Counting and analysis (Others send samples home and count in their own counters)
- The thorium limit is comparable to the best EXO measurement
- We would like to improve the Uranium limit
 - **Higher mass?** - Probably not much help. We would like to do ~5 times better, but 5 times more mass is 200 grams. Might be possible, but it will take more R&D
 - **Shorter cooling off period?** Probably our best bet. The ^{239}Np has a 2.4 day half-life. However, currently NC State uses aluminum stingers, which are very 'hot' when first removed. Requires more R&D.



NAA at NC State

(conversation with S. Lassell in Nucl. Engineering)

- Currently 1200-2000 samples per year
- Large room for expansion
 - Room for more samples
 - Run reactor longer
- Sample size (3 different vials)
 - 0.13 ml – dia. 0.22" X 0.44"
 - 1.4 ml – dia. 0.38" X 0.87"
 - 8.06 ml – dia. 0.57" X 2.09"
 - www.lacontainer.com
 - Vials supplied by NC State
- Typical irradiation – 24 hr
- Neutron E and flux can be somewhat tailored
 - Further from core or shielding = lower flux & lower energy
- Detectors
 - 3 HPGe detector
 - Relative efficiency order 20-30%
 - One automated detector system for large number of samples
 - 1 low background HPGe detector with large shield
- Estimated costs for NC State physics
 - Irradiation - \$100/hour
 - Analysis (typical including irradiation)
 - \$55 for one element
 - \$10-\$20 for second element
 - \$10/element after two
 - U and Th - \$70/sample
- Sensitivity test underway (irradiation of polyethylene)
- More sensitive gamma at Kimballton (REF: P. Finnerty talk)



Other facilities

- The University of Missouri-Columbia Research Reactor Center (MURR)
 - 10 MW research reactor
 - Offers NAA service and we can request quote online
 - <http://www.murr.missouri.edu/>
- Oak ridge – High Flux Isotope Reactor (HFIR)
 - 85 MW reactor
 - Uses both plastic and graphite vials (graphite = longer irradiation)
 - Quote from website *“The NAA systems support ORNL (DOE) programs, are used in work-for-others projects, and are available for use by students and faculty of universities through Oak Ridge Associated Universities and other programs.”* – NC State, Duke, and USC are part of the Oak Ridge Associate Universities
- MIT Nuclear Reactor Laboratory (NRL)
 - 5 MW reactor
 - 4 HPGe detectors
 - More info at - http://web.mit.edu/nrl/www/research/neutron_activation.htm

